## A Short and General Synthesis of 1,3-Dihydro-6-heteroaryl-5-perfluoroalkyl-2*H*-imidazo[4,5-*b*]pyridin-2-ones

Gee-Hong Kuo\*, Edward R. Bacon, Baldev Singh, Michael A. Eissenstat and George Y. Lesher [1]

Department of Medicinal Chemistry,
Sterling Winthrop Pharmaceuticals Research Division,
Rensselaer, NY 12144
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Trifluoro- or pentafluoroacylation of heteroaryl-enamines 2a,b gave the corresponding perfluoroacylated heteroaryl-enamines 3a-c. Heating the latter compounds with diethyl iminomalonate gave 2-amino-3-pyridinecarboxylates 4a-c. Hydrolysis to the free acids 5a-c, and reaction with diphenylphosphoryl azide afforded the desired 1,3-dihydro-6-heteroaryl-5-perfluoroalkyl-2*H*-imidazo[4,5-*b*]pyridin-2-ones 6a-c.

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The preparation of fluorinated analogues of biologically active compounds has long been of interest to the pharmaceutical and agricultural industry [2]. In support of our cardiovascular program, we needed to prepare some 6-heteroaryl-5-perfluoroalkyl imidazo[4,5-b]pyridines. While selective substituted imidazo[4,5-b]pyridines are known compounds [3,4], 6-heteroaryl-5-perfluoroalkyl imidazo-[4,5-b]pyridines have not been reported. We now report our discovery of a short and general synthesis of these novel 1,3-dihydro-6-heteroaryl-5-perfluoroalkyl-2H-imidazo-[4,5-b]pyridin-2-ones [5].

Ar 6 Ar = heteroaryl

R 5 N 3 N R = 
$$CF_3$$
 or  $C_2F_5$ 

Treatment of the commercially available 4-picoline la and 4-methylquinoline 1b with Brederick's reagent, 1-(1,1dimethylethoxy)-N,N,N',N'-tetramethyl-methanediamine, in dimethylformamide at 140° afforded the N,N-dimethyl-2-(4-pyridinyl)ethenamine 2a (80%) [6] and N,N-dimethyl-2-(4-quinolinyl)ethenamine 2b (80%) [6], respectively. Further treatment of 2a,b with trifluoroacetic or pentafluoropropionic anhydride [7] in dichloromethane in the presence of triethylamine provided the corresponding trifluoro or pentafluoroacylated enamines 3a-c in high yields (75-92%). In contrast, attempts to prepare 3a employing the procedure reported for the preparation of acetyl enamine 3 [8] (Ar = 4-pyridinyl, R = methyl) afforded 2aas the only isolated product (60%). The assignment of the E isomers 3a-c were based upon the NOE experiments of 3a and 3b. A NOE study of 3a showed that irradiation of the methyl groups enhanced both H<sub>x</sub> (17%) and Ha (8%) in agreement with the cis relationship of the dimethylamino group and the pyridine. Similarly, irradiation of the methyl groups of **3b** enhanced H<sub>v</sub> (18%), H<sub>b</sub> (6%) and H<sub>c</sub> (3%). Interestingly, the attempted acetylation

of **2a,b** with acetic anhydride was unsuccessful although the acetylation of enamines was documented [9].

Ha N(CH<sub>3</sub>)<sub>2</sub>
Hx
Hc
$$CF_3$$
 $3 \text{ a}$ 
 $3 \text{ b}$ 

Reaction of 3a-c with the easily prepared diethyl iminomalonate [10] gave the desired 2-amino-3-pyridinecarboxylates 4a-c directly, albeit in low yields (22-30%) [11]. This single step transformation avoids an alternative three-step sequence normally used for the preparation of related 2aminopyridines, i.e. conversion of the acetyl enamines to pyridinones, chlorination of pyridinones to 2-chloropyridines and a high pressure amination of 2-chloropyridines to 2-aminopyridines [12,13]. Carboxylates 4a-c were readily hydrolyzed with 10% aqueous sodium hydroxide to the 2-amino-3-pyridinecarboxylic acids 5a-c which were used for the next step directly without purification. A Curtius rearrangement [14] was effected by the reaction of 5a-c with diphenylphosphoryl azide in dioxane at 110° to give the corresponding isocyanate intermediates which underwent intramolecular cyclization with the 2-amino group to afford the desired imidazo[4,5-b]pyridin-2-ones 6a-c in good yields (42-81%) (Scheme I).

In summary, considering the easy accessibility of heteroarylenamines 2 [6] and fluorinated alkanoic anhydrides, and the mild reaction conditions, the five-step synthetic route depicted in Scheme I represents a general method for the preparation of novel 1,3-dihydro-6-heteroaryl-5-perfluoroalkyl-2*H*-imidazo[4,5-*b*]pyridin-2-ones.

#### Scheme I

1, 2	Ar	3, 4, (5), 6	Ar	R
а	4-pyridinyl 4-quinolinyl	а	4-pyridinyl 4-quinolinyl	CF <sub>3</sub> CF <sub>3</sub>
b		b		
	, ,	С	4-pyridinyl	C <sub>2</sub> F <sub>5</sub>

#### **EXPERIMENTAL**

All reagents were commercially available and used without further purification. Dichloromethane, p-dioxane and DMF were dried over 4 A molecular sieves prior to use. Melting points were obtained on a Thomas-Hoover melting point apparatus, and are uncorrected. The ir spectra were obtained using a Nicolet 20 SX spectrometer. The 'H nmr spectra were recorded on a GE QE 300 (300 MHz) spectrometer. Mass spectra were obtained using a Nermag R 10-10C spectrometer. Elemental analyses were performed by Galbraith Laboratories Inc., Knoxville, TN.

### N,N-Dimethyl-2-(4-pyridinyl)ethenamine (2a).

Compound 2a was prepared according to the literature procedure [6].

N,N-Dimethyl-2-(4-quinolinyl)ethenamine (2b).

Compound 2b was prepared according to the literature procedure [6].

4-(Dimethylamino)-1,1,1-trifluoro-3-(4-pyridinyl)-3-buten-2-one (3a). General Procedure.

A mixture of 2a (9.45 g, 63.9 mmoles) and triethylamine (12.9 g, 128 mmoles) in dichloromethane (100 ml) was cooled to 0°, and a solution of trifluoroacetic anhydride (14.7 g, 70.2 mmoles) in dichloromethane (5 ml) was added dropwise over 30 minutes. The reaction mixture was allowed to warm to 20° and stirred for 16 hours. The solvent was removed and the oily residue extracted with ether (4 x 200 ml). Silica gel (6 g) and magnesium sulfate (5 g) was added to the ether solution which was stirred and filtered. The filtrate was concentrated *in vacuo* and the residue solidified at 0° to give a yellow-orange oily solid (27 g). This crude oily solid

was used directly in the next reaction. The crude oily solid of another run starting with **2a** (4.3 g, 29 mmoles) was further purified by flash chromatography on silica gel (dichloromethane/acetone, 4:1) and then crystallized from dichloromethane/hexane to give **3a** as yellow crystals, yield 5.5 g (78%), mp 82-84°; ir (potassium bromide):  $\nu$  1591, 1410, 1177, 1126 cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.46 (br s, 3 H), 3.22 (br s, 3 H), 7.13 (d, 2 H, J = 5.8 Hz), 7.69 (s, 1 H), 8.55 (d, 2 H, J = 5.7 Hz); ms: (CI) m/z 245 (M+1, s).

Anal. Calcd. for  $C_{11}H_{11}N_2F_3O$ : C, 54.10; H, 4.54; N, 11.47; F, 23.34. Found: C, 54.02; H, 4.52; N, 11.43; F, 23.37.

4-(Dimethylamino)-1,1,1-trifluoro-3-(4-quinolinyl)-3-buten-2-one (3b).

The same procedure described for **3a** was followed for the reaction of **2b** with trifluoroacetic anhydride. The product was purified by flash chromatography on silica gel (dichloromethane/acetone, 1:2) and crystallized from dichloromethane/hexane to give **3b** as light-yellow crystals, yield 75%, mp 135-137°; ir (potassium bromide):  $\nu$  1589, 1504, 1435, 1404, 1245, 1180 cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.18 (br s, 3 H), 3.21 (br s, 3 H), 7.33 (d, 1 H, J = 4.3 Hz), 7.56 (br t, 1 H, J = 7.3 Hz), 7.73 (br t, 1 H, J = 7.7 Hz), 7.82 (d, 1 H, J = 8.1 Hz), 7.99 (s, 1 H), 8.15 (d, 1 H, J = 8.4 Hz), 8.90 (d, 1 H, J = 4.3 Hz); ms: (CI) m/z 295 (M+1, s). Anal. Calcd. for  $C_{15}H_{13}N_2OF_3$ : C, 61.22; H, 4.45; N, 9.52; F, 19.31. Found: C, 60.98; H, 4.40; N, 9.46; F, 19.35.

1-(Dimethylamino)-4,4,5,5,5-pentafluoro-2-(4-pyridinyl)-1-penten-3-one (3c).

The same procedure described for **3a** was followed for the reaction of **2a** with pentafluoropropionic anhydride. The product was purified by flash chromatography on silica gel (ether/dichloromethane, 1:1) and crystallized from ethyl acetate/hexane to give **3c** as light-yellow crystals, yield 92%, mp 105-106°; ir (potassium bromide):  $\nu$  2984, 2942, 1582, 1490, 1359, 1195 cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.44 (br s, 3 H), 3.19 (br s, 3 H), 7.13 (d, 2 H, J = 6.2 Hz), 7.73 (s, 1 H), 8.55 (d, 2 H, J = 6.2 Hz); ms: (CI) m/z 295 (M+1, s).

Anal. Calcd. for C<sub>12</sub>H<sub>11</sub>N<sub>2</sub>F<sub>5</sub>O: C, 48.99; H, 3.77; N, 9.52. Found: C, 48.69; H, 3.75; N, 9.42.

6-Amino-2-(trifluoromethyl)-[3,4'-bipyridine]-5-carboxylic Acid Ethyl Ester (4a). General Procedure.

Diethyl iminomalonate hydrochloride (20 g, 102.3 mmoles) [10] was added slowly to an ice-cold mixture of sodium bicarbonate (17 g), water (100 ml) and ether (60 ml) with rapid stirring. The ether layer was separated and the aqueous layer extracted with ether (3 x 30 ml). The combined ether solution was dried (sodium sulfate) and concentrated to give diethyl iminomalonate (16.1 g) as a colorless oil. The colorless oil was mixed with the crude oily solid 3a (20 g, 37 mmoles) and stirred at 75° for 16 hours. The reaction mixture was dissolved in water (250 ml), acidified with acetic acid and extracted with ether (3 x 100 ml), dried (sodium sulfate) and concentrated. The crude product was purified by flash chromatography on silica gel (ethyl acetate/hexane, 1:5 to 1:3) and crystallized from dichloromethane/hexane to give 4a as white crystals, yield 3.5 g (30%), mp 121-125°; ir (potassium bromide): v 3440, 3315, 3240-3100, 1711, 1638, 1600 cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  1.37 (t, 3 H, J = 7.2 Hz), 4.37 (q, 2 H, J = 7.2 Hz), 7.24 (d, 2 H, J = 4.0 Hz), 8.09 (s, 1 H), 8.66 (d, 2 H, J =3.8 Hz); ms: (CI) m/z 312 (M + 1).

Anal. Calcd. for C<sub>14</sub>H<sub>12</sub>N<sub>3</sub>O<sub>2</sub>F<sub>3</sub>: C, 54.02; H, 3.89; N, 13.50; F, 18.31. Found: C, 54.34; H, 4.07; N, 13.18; F, 18.23.

2-Amino-5-(4-quinolinyl)-6-(trifluoromethyl)-3-pyridinecarboxylic Acid Ethyl Ester (4b).

The same procedure described for **4a** was followed, but using **3b** instead of **3a**. The product was purified by flash chromatography on silica gel (ethyl acetate/hexane, 1:10 to 1:5) and crystallized from dichloromethane/hexane to give **4b** as white crystals, yield 22%, mp 204-206°; ir (potassium bromide):  $\nu$  3435, 3286, 1704, 1619, 1584, 1236 cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  1.34 (t, 3 H, J = 7.0 Hz), 4.35 (m, 2 H), 7.30 (d, 1 H, J = 4.3 Hz), 7.52 (m, 2 H), 7.77 (m, 1 H), 8.14 (s, 1 H), 8.21 (d, 1 H, J = 8.5 Hz), 8.98 (d, 1 H, J = 4.3 Hz); ms: (CI) m/z 362 (M+1, s).

Anal. Calcd. for C<sub>18</sub>H<sub>14</sub>N<sub>3</sub>O<sub>2</sub>F<sub>3</sub>; C, 59.84; H, 3.91; N, 11.63; F, 15.77. Found: C, 59.73; H, 3.90; N, 11.50; F, 15.93.

6-Amino-2-(pentafluoroethyl)-[3,4'-bipyridine]-5-carboxylic Acid Ethyl Ester (4c).

The same procedure described for 4a was followed, but using 3c instead of 3a. The product was purified by flash chromatography on silica gel (ether/hexane, 1:1) and crystallized from dichloromethane/hexane to give 4c as white crystals, yield 24%, mp  $125-127^\circ$ ; ir (potassium bromide):  $\nu$  3442, 3140-3053, 1698, 1628, 1309, 1249, 1228 cm<sup>-1</sup>; 'H nmr (deuteriochloroform):  $\delta$  1.36 (t, 3 H, J = 7.1 Hz), 4.37 (q, 2 H, J = 7.1 Hz), 7.21 (d, 2 H, J = 5.7 Hz), 8.04 (s, 1 H), 8.64 (d, 2 H, J = 5.7 Hz); ms: (CI) m/z 362 (M+1, s).

Anal. Calcd. for C<sub>15</sub>H<sub>12</sub>N<sub>3</sub>O<sub>2</sub>F<sub>5</sub>: C, 49.87; H, 3.35; N, 11.63; F, 26.29. Found: C, 49.67; H, 3.20; N, 11.53; F, 26.14.

6-Amino-2-(trifluoromethyl)-[3,4'-bipyridine]-5-carboxylic Acid (5a). General Procedure.

A mixture of **4a** (2.2 g, 7.1 mmoles), 10% aqueous sodium hydroxide (12 ml) and methanol (20 ml) was stirred at 100° for 6 hours. The reaction mixture was concentrated *in vacuo* and the aqueous mixture was neutralized with dilute acetic acid to pH 4-5. The solid product was collected, washed with water, dichloromethane and dried to give **5a** as a white solid; yield: 1.76 g (88%); mp > 283° dec; ir (potassium bromide):  $\nu$  3502, 3338, 2432, 1919, 1701 cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriotrifluoroacetic acid):  $\delta$  8.30 (d, 2 H, J = 7.1 Hz), 8.91 (s, 1 H), 9.15 (d, 2 H, J = 7.1 Hz); ms: (CI) m/z 284 (M+1).

No attempts were made to prepare an analytical sample.

1,3-Dihydro-6-(4-pyridinyl)-5-(trifluoromethyl)-2*H*-imidazo[4,5-*b*]-pyridin-2-one (**6a**). General Procedure.

A mixture of **5a** (1.5 g, 5.3 mmoles), diphenylphosphoryl azide (1.9 g, 6.9 mmoles) and triethylamine (1.5 ml) in p-dioxane (30 ml) was stirred at 110° for 6 hours. The reaction mixture was concentrated in vacuo and the residue treated with water (8 ml) and acetic acid (2 ml). The solid was collected, washed with water and air dried to give a yellowish white solid. The crude product was recrystallized from methanol to give **6a** as white crystals, yield 1.2 g (81%), mp >300°; ir (potassium bromide):  $\nu$  3600-3000, 1725, 1600, 1480 cm<sup>-1</sup>; <sup>1</sup>H nmr (acetone-d<sub>6</sub>):  $\delta$  7.29 (s, 1 H), 7.38 (d, 2 H, J = 4.9 Hz), 8.65 (d, 2 H, J = 4.9 Hz); ms: (CI) m/z 281 (M+1).

Anal. Calcd. for  $C_{12}H_7N_4OF_3$ : C, 51.44; H, 2.52; N, 19.99. Found: C, 51.56; H, 2.54; N, 19.90.

1,3-Dihydro-6-(4-quinolinyl)-5-(trifluoromethyl)-2*H*-imidazo[4,5-*b*]-pyridin-2-one (**6b**).

The same procedure described for **5a** was followed to prepare **5b** as a pale yellow solid. Compound **5b** was used directly without purification to prepare **6b** following the same procedure described for **6a**. Compound **6b** was purified by flash chromatography on silica gel (dichloromethane/methanol, 9:1) and crystallized from methanol to give **6b** as a white solid; yield 80% from **4b**, mp > 300° dec; ir (potassium bromide):  $\nu$  3110-2708, 1722, 1248, 1199 cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriotrifluoroacetic acid):  $\delta$  7.90 (d, 1 H, J = 8.7 Hz), 7.91 (s, 1 H), 8.06 (t, 1 H, J = 7.9 Hz), 8.11 (d, 1 H, J = 5.8 Hz), 8.34 (t, 1 H, J = 7.9 Hz), 8.49 (d, 1 H, J = 8.7 Hz), 9.26 (d, 1 H, J = 5.8 Hz); ms: (CI) m/z 331 (M+1).

Anal. Caled. for  $C_{16}H_9N_4OF_3$ ·0.25 $H_2O$ : C, 57.41; H, 2.86; N, 16.74. Found: C, 57.53; H, 2.77; N, 16.40.

# 1,3-Dihydro-5-(pentafluoroethyl)-6-(4-pyridinyl)-2H-imidazo[4,5-b]pyridin-2-one (**6c**).

The same procedure described for **5a** was followed to prepare **5c** as a yellow solid. Compound **5c** was used directly without purification to prepare **6c** following the same procedure described for **6a**. Compound **6c** was purified by flash chromatography on silica gel (dichloromethane/methanol, 15:1) and crystallized from DMF to give **6c** as white crystals, yield 42% from **4c**, mp > 290° (dec); ir (potassium bromide):  $\nu$  3500-3100, 1721, 1601 cm<sup>-1</sup>; <sup>1</sup>H nmr (acetone-d<sub>6</sub>):  $\delta$  7.27 (s, 1 H), 7.35 (br d, 2 H, J = 5.9 Hz), 8.63 (br d, 2 H, J = 5.9 Hz); ms: (CI) m/z 331 (M+1, s).

Anal. Calcd. for  $C_{18}H_7N_4OF_5$ : C, 47.29; H, 2.14; N, 16.97; F, 28.77. Found: C, 47.36; H, 2.21; N, 16.67; F, 28.89.

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